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DESIGNING AND DEVELOPMENT OF IBRUPROFEN FLOATING TABLET

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ABSTRACT

Study of Ibuprofen Floating tablets were to develop optimized gastric floating drug delivery system(GFDDS) by using polymers like HPMC K4M and Carbopol 940 to increase the bioavailability and therapeutic efficacy of ibuprofen. Floating drug delivery system have a bulk density lower than gastric fluids, thus remain buoyant in the stomach causing an increase in gastric residence time. The compatibility evaluations were performed by DSC analysis. Studies imply that polymers are compatible with each other. There was no interaction found between polymer and drug. The research was undertaken with the aim to formulate and characterize the sustained release floating tablets of Ibuprofen using HPMCK4M and Carbopol 940 as polymers.

KEYWORDS: Ibuprofen, buoyancy lag time, HPMCK4M, Carbopol 940.

INTRODUCTION

Oral drug delivery has been known for the most widely utilized route of administered among all the routes that have beenfor the systemic delivery of drug via various pharmaceutical products of different dosage forms. The characteristics of the dosage form are not significant for the drugs that: They Are insoluble in intestinal fluids Act locally to overcome these limitations, various approaches have been proposed to increase gastric residence of drug delivery systems in the upper part of the gastrointestinal tract includes floating drug dosage systems (FDDS) swelling or expanding systems, mucoadhesive systems, modified-shape systems, high-density system, and other delayed gastric emptying devices. Among these systems, FDDS have been most commonly used. When the dosage form administered it contact

with gastric fluid and produce effervescent and evolved CO2 gas. This support to penetrate the fluid in tablet and float, the low density polymer HPMC various grade provide low density system so it buy out efficiently in gastric fluid.

MATERIALS AND METHOD

Ibuprofen obtained as a gift sample from Savan Pharma Pvt Ltd., Hyderabad, Carbopol 940 and HPMC K4M was supplied from Bio - Gen extracts Pvt Ltd., Bangalore, Citric acid, lactose and Sodium bicarbonate were obtained from Qualigens Fine Chemicals, Mumbai, India, Talc and Microcrystalline cellulose were obtained from Lobachemiepvt., ltd., Mumbai, India. All the chemicals and reagents required for the present experimental work are of analytical grade.

METHOD OF PREPARATION OF IBUPROFEN FLOATING TABLETS

The ibuprofen floating tablets were prepared by blending the drug (ibuprofen), polymer s(HPMCK4M) and Carbopol940 in different proportions respectively. To this sodium bicarbonate, lactose, citric acid were added to mortar and pestle according to their geometric dilution and finally make up the total weight of tablet using micro crystalline cellulose. The powder was passed through sieve no.60. The obtained samples were collected and re triturated. To this required amount of talc is added and compressed finally. In the present work, 4 formulations (F1 to F4) floating tablets of Ibuprofen were prepared using variable concentrations of HPMCE5M and Carbopol940 (Table No.01).

IN – VITRO CHARACTERIZATION

• Weight uniformity test

If the drug forms greater part of the tablet, any variation in the tablet weight obviously indicates a variation in the active ingredient this test resembles weight uniformity test. 20 tablets were selected at random and average weights were determined. Then individual tablets weighed and the individual weight was compared with the average.

Calculate the average weight of tablets =
$$\frac{Total\ weight\ of\ tablets}{Number\ of\ tablets}$$

Average weight of tablets (X) = (X1+X2+X3+...+X20) / 20

• Hardness uniformity studies

The hardness of prepared formulation was measured by using Pfizer hardness tester. Five

floating tablets were used for hardness uniformity studies. The hardness data used to calculate mean and standard deviation.

Thickness uniformity studies

The thickness uniformity studies were carried out by using Vernier calipers. Five tablets were used for thickness uniformity studies and denoted in millimeter. The data obtained was used to calculate mean and standard deviation.

• Friability (F)

The friability of the tablet was determined using Roche Friabilator. It is expressed in percentage (%). 20 tablets were initially weighed (W initial) and transferred into the friabilator. The friabilator was operated at 25 rpm per min for 4 mins (100 revolutions).

The tablets were weighed again (W final). The % friability was then calculated by

$$F = \frac{W \ 1initial - W2 \ final}{W1 \ initial} \times 100$$

· Thickness and diameter

Tablet thickness is important for tablet packaging; very thick tablets affect packaging either. in blisters or plastic containers. The tablet thickness is determined by the Diameter of the die, the amount of fill Permitted to enter the die and the force or Pressure applied during compression. The Thickness of the tablet may be measured Manually or by automatic equipment. The Thickness and diameter of the tablets was Measured by Vernier Calipers. It is expressed In mm.

Content uniformity

Twenty tablets were taken and amount of Drug present in each tablet was determined. The tablets were crushed in a mortar and the Powder equivalent to 100mg of drug was Transferred to 100ml standard flask. The Powder was dissolved in a suitable solvent And make up the final volume with the Suitable buffer solution. The sample was Mixed thoroughly and filtered through a 0.45μ membrane filter. The filtered solution Was diluted suitably and analyzed for drug Content by UV spectrophotometer, using Buffer solution as a blank.

In vitro buoyancy / floating study

In vitro buoyancy studies were performed for All the formulations. The randomly selected

Tablets from each formulation were kept in a 200ml beaker containing simulated gastric Fluid, pH 1.2. The time taken for The tablet to rise to the surface and float was Taken as floating lag time (FLT). The duration Of time the dosage form constantly remained On the surface of medium was determined as The total floating time (TFT).

Swelling Index

The Swelling index of tablets was found by Placing the tablets in the basket of dissolution Apparatus using dissolution medium pH 6.8 Buffer at 37 ± 0.50 C. After 0.5, one, two, three, Four, five, six, seven and eight hours, each Dissolution basket containing tablet was Withdrawn and blotted with tissue paper to Remove the excess water and weighed on the Analytical balance. The experiment was performed in triplicate for Each time point. Tablets were randomly selected and one Tablet was introduced in each tube Disintegration apparatus and placed in 1 litre Beaker containing water at 370 ± 20 C and the Time of disintegration was recorded.

In vitro dissolution studies

The release rate of aceclofenac from floating Tablets was determined using United States Pharmacopeia (USP) Dissolution Testing Apparatus 2 (paddle method). The Dissolution test was performed using 900 ml Of pH 1.2 HCL buffer for 2 hrs followed by pH 6.8 Phosphate buffer for 8hrs. A sample (5 ml) of the solution was withdrawn from the dissolution apparatus hourly and the samples were replaced with fresh dissolution medium. The samples were filtered through a 0.45μ Membrane filter and diluted to a suitable Concentration with of pH 1.2 HCL buffer for 2 Hrs followed by pH 6.8 Phosphate buffer for 8hrs. Absorbance of these solutions was Measured at 222 nm using a UV/Visible Spectrophotometer.

Drug release kinetics

The success of HPMC K4M with Carbopol940 In controlling the release of the drug was Studied under the following heads to Understand the order and probable Underlying mechanism involved in the Release pattern.

DSC

DSC curves: (A) ibuprofen, (B) physical Mixture. DSC thermo gram of pure ibuprofen Exhibited a single endothermic response Corresponding to the melting of the ibuprofen At 77.5°C. As mixture did not show any fusion Peak or phase transition, apart from a broad Endotherm, as a result of dehydration, which lies between 80°C and 120°C. For the PM, the

Endotherm broadened and was shifted Slightly to a lower temperature (76.8°C), Reflecting a partials change of ibuprofen Crystal structure.

RESULTS AND DISCUSSION

In the preparation of gastro-retentive drug Delivery systems. These include floating Systems, swell able and expandable systems, High density systems, bioadhesive systems, Altered shape systems, gel forming solution or Suspension systems and sachet systems. Various approaches have been followed to Encourage gastric retention of an oral dosage Form. Floating systems have low bulk density So that they can float on the gastric juice in The stomach.

The prepared Ibuprofen FDDS were subjected To various evaluation studies done like Weight Uniformity test, Hardness, Thickness, Friability (F)Thickness, Content uniformity, In vitro Buoyancy / floating study, Swelling Index Disintegration studies, In vitro dissolution Studies.

Evaluation of tabletsWeight Variation, Thickness, Hardness and Friability

The results showed that weight variation, thickness were lying within limits. Because of variation in the compressional forces there is a slight variation in hardness of tablets. As the proportion of polymers increases the hardness of the tablets was found to increase in case of HPMC. The friability loss was found to be within the limits in all the friability tablet was found to mechanically strong.

Buoyancy and total Flotation test

From the results, it was observed that as the buoyancy lag time and the total floating time was studied for all the formulations. F1, F2, F3 and F4 total floating time were found to be 12, 10, 5.5 and 7 hrs respectively as shown in Table No. 3. F1 showed optimum buoyancy lag time. for all the F1 and F2 formulations showed more total floating time when compared to F3 and F4 due to the presence of hydrophobic polymer which decreased the solubility. When compared in between F2 and F4, F4 showed less total floating time. Thus with an increase in the concentration of the hydrophilic polymer total floating time was found to be decreased due to increase in the solubility. Results revealed that as the concentration of the hydrophilic polymer increases, the buoyancy lagging time decreases. The increase in the concentration of the hydrophobic polymer resulted in the increase of the buoyancy lag time. Thus polymer HPMCK4M and Carbopol 940 were found to have optimum floating characters for a longer period.

Swelling thickness

The extent of swelling was found out by measuring the thickness of the tablet before and after one hour stay of the tablet in pH 6.8 buffer at 37 ± 0.5 0C. FormulationF1 And F2 tablets were found to swell more and formulationF3 and F4 tablets were swelling to lesser extent.

Drug content

Drug content of all the formulations are within the acceptable range which shows the proper mixing of the drug with excipients as shown in Table No. 04.

In-vitro drug release

Formulations with polymers showed high less which retards the drug release to a greater extent. Thus the HPMC Decreasing and Carbopol 940 Increasing concentration provide the optimum drug release.

Drug release kinetics

From the data of drug releasefound all the tablet formulations follow diffusion mechanism for drug release. The Higuchi square root equation describes the release from systems where the solid drug is dispersed in an insoluble matrix and the rate of drug release is related to the rate of drug diffusion.

DIFFERENTIAL SCANNING CALORIMETRY (DSC) ANALYSIS

In order to investigate the possible interactions between the drug and polymers used, differential scanning calorimetric studies were carried out. DSC thermo gram of the formulation was compared with the DSC thermo gram of the pure drug. The pure ibuprofen displayed a sharp endothermic peak at 820C corresponding to the melting point of the drug and a similar peak was also observed in the formulation.

Table 1: Development of different formulations containing, varying proportions of polymers.

Formulation	Drug	HPMC	Carbopol	NaHCo3	M.C.C	Citric	Lactose	Mg stearate	Talc
(mg)	(mg)	(mg)	(mg)	(mg)	(mg)	acid (mg)	(mg)	(mg)	(mg)
F1	200	100	25	50	75	25	25	5	10
F2	200	75	50	50	75	25	25	5	10
F3	200	50	75	50	75	25	25	5	10
F4	200	25	100	50	75	25	25	5	10

Table 2: Weight Variation, Thickness, Hardness and Friability.

Formula	Weight variations	Thickness	Hardness	Friability
F1	0.239	0.38 ± 0.031	5.6	0.3
F2	0.247.	0.40 ± 0.011	6.0	0.5
F3	0.236	0.41±0.007	4.0	0.8
F4.	O.246.	0.43 ± 0.007 .	3.6.	0.7

Table 3.

Formulation	Buoyancy lag	Total Floatation time	
code	time (min)	(hrs)	
F1	2.5 min	13	
F2	2.0 min	11	
F3	4.0 min	5.6	
F4	4.5 min	8	

Table 4: Data showing the drug Content of various formulations of Ibuprofen.

Formulation code.	% Drug content
F1	98
F2	97
F3	93
F4	94

Table 5: Standard curve of lbuprofen.

S.No	Concentration(µg/ml)	Absorbance
1	10	0.341
2	20	0.749
3	30	1.123
4	40	1.487
5	50	1.875

Table 6: Data showing comparative In-Vitro % drug release profiles for all the prepared formulations.

Time(Hrs)	F1	F2	F3	F4
30(Mins)	1.01	2.37	2.98	4.26
1	1.08	6.49	6.31	7.25
2	3.62	7.63	8 15	9.08
3	9.97	9.15	10.0	10.98
4	13.67	11.06	11.98	12.0
5	15.61	13.87	13.97	13.93
6	18.48	15.82	20.38	18.58
7	21.38	16.89	23.3	22.38
8	22.52	19.77	30.76	30.72
9	25.45	25.39	37.38	37.35
10	27.52	29.25	48.58	47.87
11	31.42	36.76	51.9	54.49
12	35.36	43.45	70.35	75.77
13	40.45	54.72	80.1	90.98

SUMMARY

In present work an attempt has been made to formulate sustained release floating drug delivery system of Ibuprofen tablets by using polymers. FDDS were prepared using polymers of HPMC K4M and Carbopol 940 by direct compression method. Ibuprofen meets all the ideal characteristics to formulate in the form a floating drug delivery system. The compatibility evaluations were performed by DSC analysis. There was no interaction found between polymer and drug. All the formulations were characterized on the basis of their evaluation studies and invitro dissolution studies. The compatibility evaluations were performed by DSC analysis. Studies imply that polymers are compatible with each other. There was no interaction found between polymer and drug. The mechanism of drugs release from tablets was dissolution followed by USP Type-II. Two formulations F1 and F2 were able to release drug up to 13hrs and F3 and F4 released maximum drug F1 only compiles with the USP type-II, which maintained the release pattern as per mention. Rests of the formulations were unable to maintain release rate as per USP-typeII.

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