

**FORMULATION AND *IN VITRO* EVALUATION OF CONTROLLED RELEASE MEBEVERINE HYDROCHLORIDE MICROSPHERES****G.Velrajan\*<sup>1</sup>, Dr.R.Suthakaran<sup>1</sup> and V.Balasubramaniam<sup>2</sup>**<sup>1</sup>Dept of Industrial Pharmacy, Vijaya College of Pharmacy, Hayathnagar, Hyderabad.<sup>2</sup>Dept of Pharmaceutics, Saraswathi College of Pharmaceutical Sciences, Moinabad, Hyderabad.

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**\*Correspondence for  
Author**

**G.Velrajan**

Dept of Industrial  
Pharmacy, Vijaya College  
of Pharmacy, Hayathnagar,  
Hyderabad, India.

**ABSTRACT**

Microspheres of mebeverine hydrochloride, which treats irritable bowel syndrome. Mebeverine hydrochloride is highly water soluble drug, having poor oral bioavailability of drug. Mebeverine hydrochloride is known to suffer from first pass effect. It is an attempt to improve oral bioavailability, reduce the frequency of drug administration, and possibility to restrict its absorption only to the colon. In this study, microspheres of mebeverine hydrochloride was prepared by solvent evaporation techniques using Eudragit FS 30D, and Ethylcellulose as polymers and yield, particle size, encapsulation efficiencies and in vitro release of the prepared microspheres were evaluated. The results showed that percentage yield, encapsulation

efficiencies and particle size were influenced mainly by polymer concentration, type of polymer and stirring speed. From the results of the in vitro study shows that the desired release rate is achieved by FEC 4 and FER 11 formulations are releasing the drug up to 12 hrs. DSC results showing there is no interaction between drug and polymers. SEM results of optimized microspheres showing discrete, spherical microspheres.

**KEYWORDS:** Mebeverine hydrochloride, ethylcellulose N 50, eudragit FS 30 D, Solvent evaporation technique.

**INTRODUCTION**

Microsphere is defined as “a monolithic structure with the drug or therapeutic agent distributed throughout the matrix either as a molecular dispersion or as a dispersion of particles, falling in the size range 1-500  $\mu^1$ . Insoluble drug carriers for prolonged and

controlled delivery of therapeutic agents in biological system recently have generated interest. Microspheres belong to the same carrier system<sup>2</sup>. The oral route of drug administration is the most convenient and important method of administering drugs for systemic effect. Nearly 50% of the drug delivery systems available in the market are oral D.D.S. and these systems have more advantages due to patient acceptance and ease of administration. During the last decade there has been interest in developing site specific formulations for targeting drug to the colon. Colonic drug delivery has gained increased importance not just for the delivery of the drugs for the treatment of local diseases<sup>3</sup>. Diseases associated with the colon like Crohn's disease, ulcerative colitis, irritable bowel syndrome. There are various methods or techniques through which colon drug targeting can be achieved, for example, formation of prodrug, coating with pH sensitive polymers, coating with biodegradable polymers, designing formulations using polysaccharides, timed released systems, pressure-controlled drug delivery systems, osmotic pressure controlled systems. Coating of the drugs with pH-sensitive polymers provide simple approach for colon specific drug delivery<sup>3</sup>.

## MATERIALS AND METHODS

Mebeverine Hydrochloride, USP Racheem Pvt Ltd, Hyderabad. Ethyl cellulose N50 Liquid paraffin AR SD Fine Chemicals, Mumbai Span- 80 IP Petroleum ether AR Ethanol AR Eudragit FS 30D M/s Rohm pharma, Germany Whatman Filter paper Whatman® Schleicher & Schuell.

### (A) Formulation of Mebeverine Hydrochloride Microspheres<sup>1</sup>

The drug was dissolved in polymer solution made by dissolving polymer in 3:1 mixture of acetone and ethanol. The above slurry was slowly introduced into 100 ml of light liquid paraffin containing span 80 (0.5%) as surfactant while being stirred at 2200 rpm by a mechanical stirrer equipped with a three bladed propeller at room temperature. The stirring was continued for three-and-a half (3½) to four (4) hours to allow the solvents (acetone, ethanol) to evaporate completely and the formed microspheres were collected by filtration. The microspheres were washed repeatedly with n-Hexane and petroleum ether until free from oil. The collected microspheres were dried at room temperature for 24 hours. The schematic representation of formulation of eudragit microspheres of mebeverine hydrochloride was given in Table no.1

**Table 1 Different Polymers drug ratio formulations**

Formulation code(Ethyl cellulose N50)	Ratios (dg:EC)	Speed (rpm)	Formulation code(Eudragit FS 30D)	Ratios (dg:ER)	Speed (rpm)
FEC 1	1:1	2000	FED 7	1:1	2000
FEC 2	1:2	2000	FED 8	1:2	2000
FEC 3	1:3	2000	FED 9	1:3	2000
FEC 4	1:4	2000	FED 10	1:4	2000
FEC 5	1:5	2000	FED 11	1:5	2000
FEC 6	1:6	2000	FED 12	1:6	2000

**(B) Yield of microspheres<sup>4,5</sup>**

The yield of microspheres was calculated from the amount of microspheres obtained divided by the total amount of all non-volatile components

$$\% \text{ Yield} = \frac{\text{Actual weight of the microspheres}}{\text{Total weight of all non-volatile components}} \times 100$$

**(C) Particle size analysis<sup>6</sup>**

The particle size of the microspheres was measured by optical microscopy. The eye piece micrometer was calibrated using a stage micrometer and the calibration factor was used further in the calculation of the size of microspheres. The microspheres were finely spread over a slide and visualized under an optical microscope using an eyepiece micrometer. About 50 readings were taken at random and the mean  $\pm$  standard deviation was calculated. The shape of the microspheres was visualized and the photographs were taken with the aid of a binocular microscope (Quasmo, India, model PZRM 700).

**(D) Drug entrapment efficiency (DEE)<sup>4</sup>**

The amount of drug entrapped was estimated by crushing 50 mg of microspheres using mortar and pestle, and extracting drug with aliquots of 6.8 pH buffer repeatedly. The extract was transferred to a 100 ml volumetric flask and the volume was made up using 6.8 pH buffer. The solution was taken in a beaker and sonicated in a bath sonicator for 2 hours. The solution was filtered and absorbance was measured after suitable dilutions spectrophotometrically at 263 nm against an appropriate blank. The amount of drug entrapped in the microspheres was calculated using the following formula

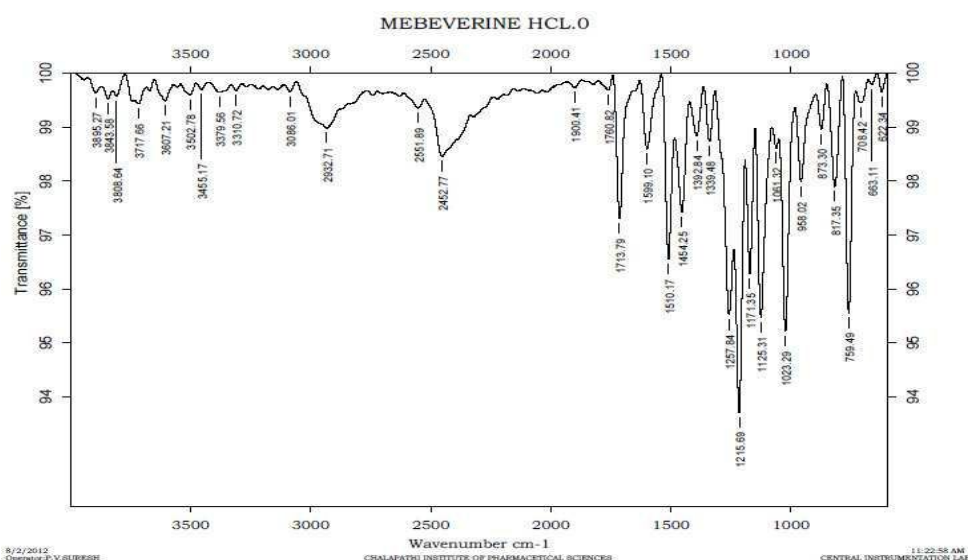
$$\text{DEE} = \frac{\text{Amount of drug actually present}}{\text{Theoretical drug load expected}} \times 100$$

**(E) *In vitro* drug release study<sup>7</sup>**

In vitro drug release studies were carried out for all formulations in USP type II dissolution test apparatus (TDT 06P, Electrolab, India). Microspheres equivalent to 135 mg of mebeverine hydrochloride taken and Studies were carried out using USP dissolution apparatus II, microspheres were placed In muslin cloth and tied to paddle temperature maintained at  $37 \pm 0.5^\circ\text{C}$ . The dissolution medium, stirred at 50 r.p.m., consisted of 900 ml 0.1N HCl for 2hours followed by buffer of pH 7.4 for 2hrs and followed by pH 6.8 up to 12hours. Samples (1ml) were withdrawn at predetermined time intervals, compensated with fresh dissolution medium and assayed spectro photometrically at 263 nm in 0.1N HCl at pH 6.8, 7.4. No interference occurred excipients at this wavelength. This solution was analysed for the drug content spectro photometrically using UV Visible spectrophotometer (elico India) at 263 nm against an appropriate blank. Three trials were carried out for all formulations. From this cumulative percentage drug release was calculated and plotted against function of time to study the pattern of drug release. The results are presented in tables and figures.

**(F) Surface morphology of the microspheres<sup>8</sup>**

The surface morphology of the targeted and controlled release microspheres was studied with the aid of a Scanning Electron Microscope (SEM).

**RESULTS****Compatibility Studies IR Spectroscopy**

**Figure 1 FT-IR graph of pure drug**

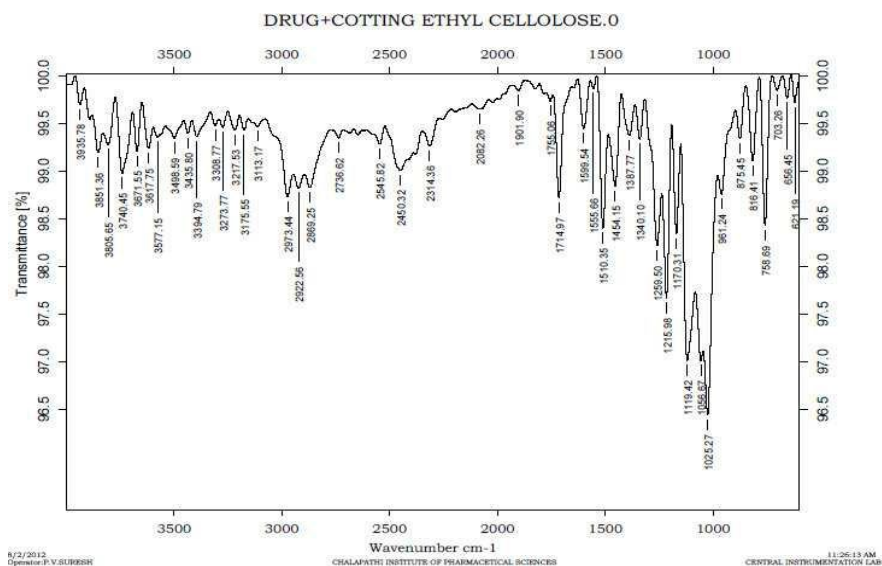


Figure 2 FT-IR graph for drug and Ethylcellulose polymer

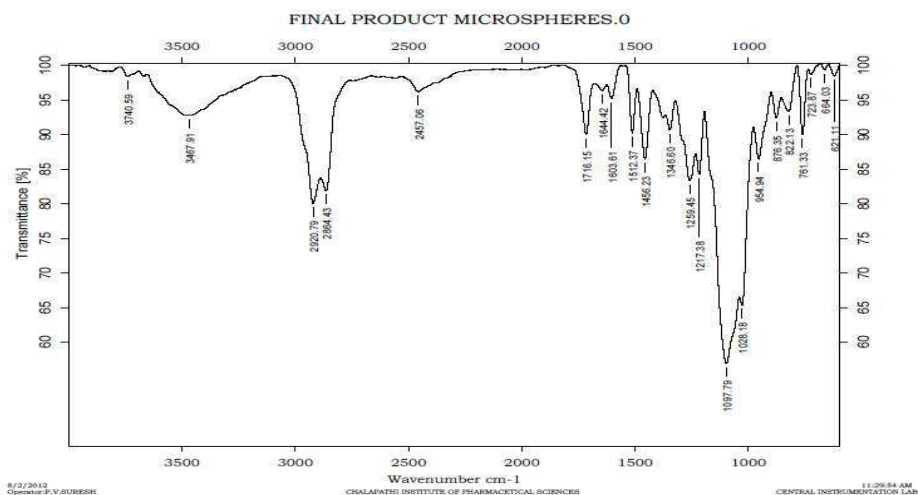


Figure 3 FT-IR graph for ethylcellulose microspheres.

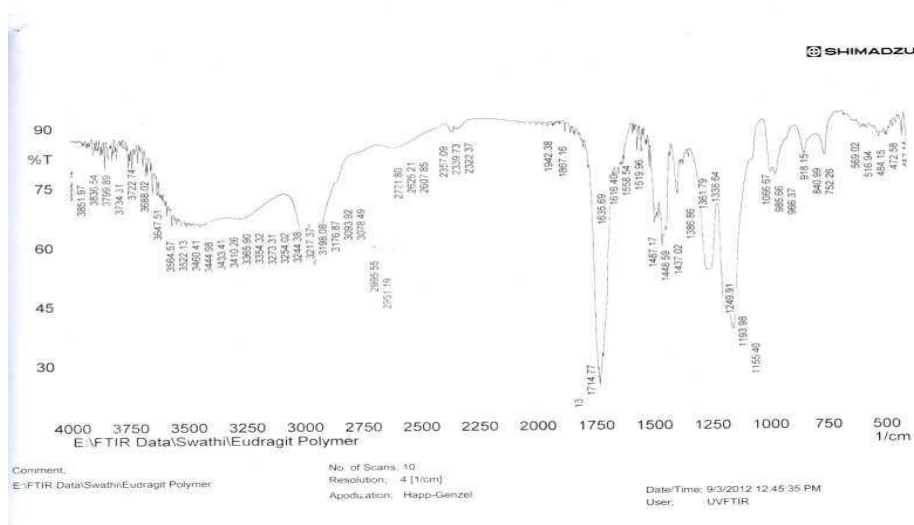
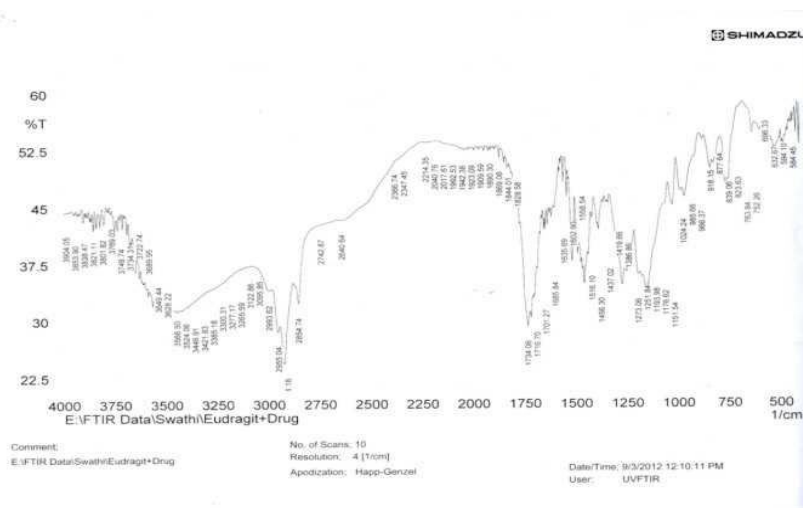


Figure 4 FT-IR graph for Eudragit polymer



**Figure 5 FT-IR grph for Eudragit final product**

**Table 2 percentage yield of all formulations**

Formulation code(Ethyl cellulose N50)	% Yield	Formulation code (Eudragit FS 30D)	% Yield
FEC 1	55.6	FED 7	45.5
FEC 2	59.3	FED 8	56.6
FEC 3	70.4	FED 9	69.3
FEC 4	79.5	FED 10	73.2
FEC 5	86.5	FED 11	86.5
FEC 6	92.2	FED 12	74.2

**Table 3 Particle size analysis of the formulation**

Formulation code (Ethyl cellulose)	Particle size (in microns)	Formulation code (Eudragit)	Particle size in microns
FEC 1 (1:1)	15.66	FED 7 (1:1)	9.12
FEC 2 (1:2)	16.44	FED 8 (1:2)	10.34
FEC 3 (1:3)	18.33	FED 9 (1:3)	11.98
FEC 4 (1:4)	19.22	FED 10 (1:4)	12.54
FEC 5 (1:5)	25.44	FED 11 (1:5)	13.43
FEC 6 (1:6)	28.77	FED 12 (1:6)	15.33

The formulated microspheres were evaluated for the size with the aid of an optical microscope. From table.3 the mean particle size of the ethyl cellulose microspheres at increasing ethyl cellulose concentrations (i.e., at drug-polymer ratios 1:1 to 1:6) increased from 15.66 to 28.77 $\mu$ m at 2200 rpm. This increase in particle size of the microspheres can be

due to an increase in viscosity with increasing polymer concentrations, which resulted in larger emulsion droplets leads to the formation of greater microspheres. The mean particle size of the Eudragit microspheres at increasing Eudragit concentrations (i.e., at drug-polymer ratios 1:1 to 1:6) increased from 9.12 to 15.33  $\mu\text{m}$  at 2200 rpm respectively.

**Table 4 Entrapment efficiency of all formulations**

Formulation code (Ethyl cellulose)	Entrapment efficiency (%)	Formulation code (Eudragit)	Entrapment efficiency (%)
FEC 1 (1:1)	70.00	FED 7 (1:1)	69.43
FEC 2 (1:2)	78.44	FED 8 (1:2)	75.33
FEC 3 (1:3)	80.32	FED 9 (1:3)	79.22
FEC 4 (1:4)	85.96	FED 10 (1:4)	84.32
FEC 5 (1:5)	89.54	FED 11 (1:5)	87.99
FEC 6 (1:6)	92.34	FED 12 (1:6)	93.22

Percentage drug entrapment efficiencies of microspheres by using different type polymers were shown in table 4. Eudragit polymeric microspheres having 93.22% (FED 12) as higher Percentage of drug entrapment efficiency for 1:6 ratio of drug to polymer. But ethyl cellulose microspheres having around 92.34% (FEC 6) as higher percentage drug entrapment efficiency for 1:6 ratio of drug to polymer, and by increasing the ratio percentage drug entrapment efficiency value is gradually increasing. This may be attributed by increase in viscosity of polymeric solution with respectively by increasing drug polymer ratio for Eudragit and ethyl cellulose polymers.

#### ***In vitro* dissolution studies of mebeverine hydrochloride microspheres**

The results of the *in vitro* drug release studies were shown in the figure 1. From the obtained dissolution data following inferences were made. On increasing the drug to polymer ratio, the drug release could be prolonged. The 1:4 ratio of drug to Eudragit FS 30D at 2000 rpm speed could sustain the release for 12 hours releasing about 82.6% of the drug. The difference in the release from the polymers may be attributed to their permeability characteristics. The available literature suggests that the quaternary ammonium groups in FS 30D are 8.85 - 11.96 % and respectively and they have a profound influence on the permeability of the polymer. Drug release from 1:4 ratio of Drug: Ethyl cellulose N50(FEC 4) showing controlled release of drug from the microspheres about 12 hours. Therefore, a high core to coat was essential to prolong the release of the drug from microspheres While increasing core to coat ratios of



drug and polymers the cumulative % drug release was decreased it was 83.47% for FED and 85.72% for FEC.

### SEM photograph of Ethylcellulose N50microspheres

Figure no.2 shows SEM photograph of optimized Ethylcellulose microspheres. SEM photographs showed discrete, spherical and uniform microspheres with smooth surface, high entrapment efficiency when compared to Eudragit microspheres, because of higher viscosity of Ethylcellulose polymer.

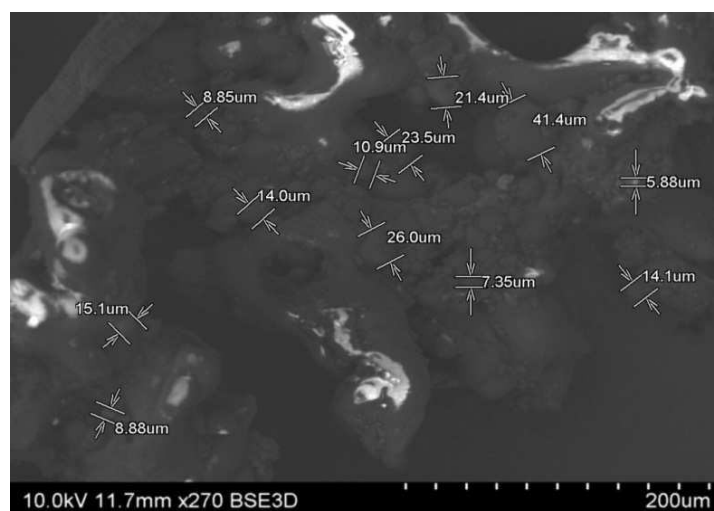


Figure 2 SEM photograph of Ethylcellulose N50microspheres

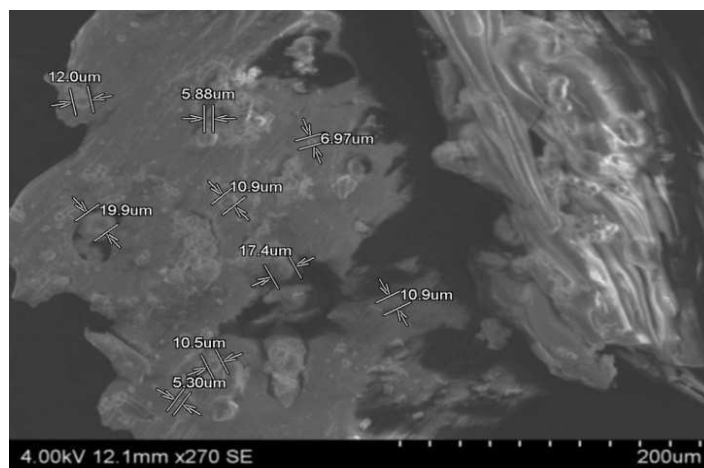


Figure 3 SEM photograph of Eudragit FS 30D microspheres

### DISCUSSIONS

Mebeverine hydrochloride is known to suffer from first pass effect. It's an attempt to improve oral bioavailability and possibility to restrict its absorption only to the colon, mebeverine



hydrochloride microspheres are prepared by solvent evaporation method. By using polymers like Ethylcellulose N50, Eudragit FS30D. Different formulations were prepared with various ratios to deliver the drug at targeted site. Mebeverine hydrochloride having short biological half life (2-3hrs) it has been selected as an ideal drug for design of oral controlled and targeted drug delivery. All formulations were formulated with various proportions, all the formulations were subjected to various evaluation tests. FT-IR, % yield, micrometric properties, entrapment efficiency, invitro release studies, DSC had shown satisfactory results. FT-IR spectra revealed that there is no interaction between polymer and drug mebeverine hydrochloride. Both the polymers are compatible. On the basis of the release data, there is increase in release time with increase in polymer ratio. Based up on evaluation parameters best formulation has been subjected to stability studies as per ICH guidelines. Formulation subjected for stability studies have been checked for drug entrapment efficiency, physical appearance up to 90days at 25°C/ 60% RH and 40°C/ 75% RH. The formulations were found to be stable because there was no significant change has been observed in the various evaluated parameters of the formulation.

## CONCLUSION

Based on results obtained it is concluded that the yield and entrapment efficiency was good for Ethyl cellulose microspheres FEC5 and eudragit microspheres FER 11 Particle size, entrapment efficiency and production yield were influenced by the type of polymer, polymer concentration, stirring speed and SEM photographs of optimized formulations FEC 5, FER 11 showed discrete, spherical microspheres. FER 11 formulations are showing good micrometric properties more targeted and controlled release than FEC 5. Therefore FER 11 formulation may be used for reducing dosing frequency thereby improving the effectiveness of drug.

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