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Research Article

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OF SOME THIENO PYRIDINE DERIVATIVES

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ABSTRACT

In the present work biodynamic activities of pyridine and its derivatives have been observed intensively. Pyridine derivatives have also been synthesized by nucleophilic substitution reactions. The constitution of the synthesized products have been characterized by using Elemental analysis, IR, ¹H NMR, ¹³C and MS. Purity of all compounds have been observed by TLC. All the newly synthesized compounds were screened for antibacterial and antifungal activities (MIC) in vitro by both dilution method with two Gram positive bacteria (S. Aureus MTCC 96 and S. pyogenus MTCC 442), two Gram negative bacteria (E. Coli MTCC 443 and P. aeruginosa MTCC 1688) and fungi A. niger MTCC 282.

KEYWORDS: Synthesis, Characterization, Biological Evaluation, Pyridine Derivatives.

INTRODUCTION

Keeping in view of wide spectrum biodynamic activities¹⁻⁴⁵ of pyridine and with a view to have therapeutic agents, the synthesis of 5 - (4 - Substituted phenoxy - 3- Nitrobenzene - 1 - Sulfonyl) - 4, 5, 6, 7 -Tetrahydrothieno[3, 2 - C] Pyridine (2a-d) have been by the nucleophilic substitution of the chloro atom of 5 - (4 - Choro - 3- Nitrobenzene - 1 - Sulfonyl) - 4, 5, 6, 7 -Tetrahydrothieno[3, 2 - C] Pyridine with different substituted phenols.

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Substituted phenols

The constitution of the synthesized products (2a-d) have been characterized by using elemental analysis, infrared and ¹H nuclear magnetic resonance spectroscopy and further supported by mass spectrometry. Purity of all compounds has been checked by thin layer chromatography.

All the newly synthesized compounds 3a-d were screened for antibacterial and antifungal activity (MIC) *in vitro* by broth dilution method⁶⁴ with two Gram positive bacteria (S. Aureus MTCC 96 and S. pyogenus MTCC 442), two Gram-negative bacteria (E. Coli MTCC 443 and P. aeruginosa MTCC 1688) and fungi A. niger MTCC 282.

Reaction scheme

R=Substituted Phenols

EXPERIMENTAL

Preparation, characterization and antimicrobial evaluation of 5 –[4- (4 - Substituted phenoxy - 3- Nitrobenzene - 1 - Sulfonyl) - 4, 5, 6, 7 –Tetrahydrothieno [3, 2 - C] Pyridine.

[A] Synthesis of scaffold-A

A mixture of 4, 5, 6, 7 –Tetra hydro thieno [3, 2 – c] pyridine hydrochloride (1.0 mole) in THF and K_2CO_3 (3.0 mole) in RBF was added at $25-30\,^{\circ}C$ for 10-15 min on magnetic stirrer. Then 4 – chloro 3 – nitro benzene sulfonyl chloride (1.1 mole) slowly added in reaction mixture with constant stirring for 15 mins at 10-15 $^{\circ}C$ and then allowed to warm at room temperature. Reaction mixture was stirred at room temperature for 2 h. After completion of reaction, the reaction mixture was distilled out. Water was added to the obtained residue and then again it stirred for 1 h 30 min. Solid was filtered out, triturated with ethanol and filtered to get pure product. The solid was dried under reduced pressure. Yield; 55.50 %, m.p. 230-232 $^{\circ}C$.

[B] Synthesis of 5 - {4 - [(4 - methoxyphenyl) sulfanyl] - 3 - nitrobenzene - 1 - sulfonyl} - 4, 5, 6, 7 - tetrahydrothieno [3, 2 - c] pyridine.

To the stirred solution of scaffold-A (1 mole) in DMF was added K_2CO_3 and 4-methoxyphenol (2 mole) at 60 0 C for 4 hours with help of magnetic stirrer. The reaction mixture was stirred at reflux temperature for 3 hours at room temperature and the solvent was evaporated under reduced pressure. The pH of the mixture was more than 7, so add dil.HCl to reduced pH upto 6.5 and again stir reaction mixture for 2 hrs. at room temperature. Water was added to the obtained residue and stirred again for 1 hour at room temperature. Solid was filtered out, triturated with ethanol and filtered to get product. The solid was dried under reduced pressure. Yield; 65%, m.p 185^{0} C. Elemental analysis; Calculated; C (53.80%), H (4.06%), N (6.27%), O (21.50%), S(14.36%). Elemental analysis; Found; C (53.78%), H(4.05%), N (6.29%), O (21.48%), S(14.35%).

Similarly, other compounds (2a-d) were synthesized by above mentioned process (B) from Scaffold-A using different substituted phenols. The physical data are recorded in Table-1.

Physical Properties

Table 1: Physical constants of $5 - \{4 - [(4 - \text{methoxyphenyl}) \text{ sulfanyl}] - 3 - \text{nitrobenzene} - 1 - \text{sulfonyl}\} - 4, 5, 6, 7 - \text{tetrahydrothieno}[3, 2 - c] pyridine.$

Spectral studies

Spectral studies like IR, ¹H NMR, ¹³C NMR, Mass Spectra (MS) and HRMS have been carried out for all the Pyridine derivatives.

Antibacterial and antifungal activity

All the newly synthesized compounds 1a-d were screened for antibacterial and antifungal ac-

tivity (MIC) *in vitro* by broth dilution method with two Gram positive bacteria (S. aureus MTCC 96 and S. pyogenus MTCC 442), two Gram-negative bacteria (E. coli MTCC 443 and P. aeruginosa MTCC 1688) and fungi A. niger MTCC 282 taking ampicillin, chloramphenicol, nystatin and griseofulvin as standard drugs. The standard strains were procured from the Microbial Type Culture Collection (MTCC) and Gene Bank, Institute of Microbial Technology, Chandigarh, India. The observed data of synthesized compounds are given in Table 2. All the glass apparatus used were sterilized before use.

RESULTS AND DISCUSSION

Table 1: Physical Data Of Compound 2a-D.

Comp.	Substitution R	Molecular Formu-	M. P. ⁰ C	Yeild %	% Composition Calculated/Found		
		la/Weight			C	H	N
2a	4-methoxy	$C_{20}H_{18}N_2S_2O_6$	185	65	53.80	4.06	6.27
	phenol	446			53.78	4.05	6.29
2b	4-hydroxy	$C_{20}H_{15}N_3S_2O_5$	193	65	54.41	3.42	9.52
	benzonitrile	441			54.40	3.39	6.50
2c	2-chloro	$C_{19}H_{15}ClN_2S_2O_5$	172	68	50.61	3.35	6.21
	phenol	451			50.59	3.33	6.19
2d	4-fluoro	$C_{19}H_{15}FN_2S_2O_5$	178	68	52.53	3.48	6.45
	phenol	434			52.51	3.46	6.43

Spectral studies

IR Spectral study of 5-{4-[(4-methoxyphenyl)sulfanyl]-3-nitrobenzene-1-sulfonyl}-4,5,6,7-tetrahydrothieno[3,2-*c*]pyridine (**2-a**).

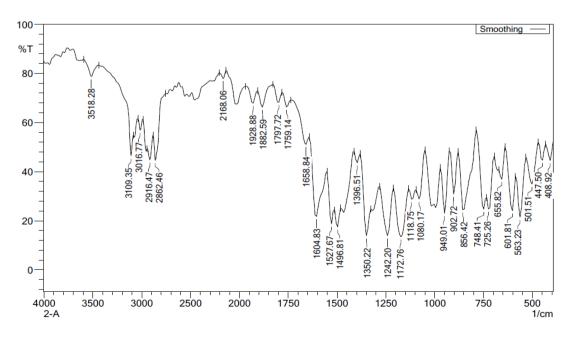


Figure 1

Table 2: IR data of compound 2a.

Vibrating Frequen- cy (cm ⁻¹)	Functional group		
3109, 3016	Aromatic C-H stretching		
2916, 2862	Aliphatic C-H Stretching		
1604	C=C stretching		
1527	NO ₂ Stretching		
1350, 1242	S=O stretching		

¹H NMR Spectral study of 5 - {4 - [(4 - methoxyphenyl) sulfanyl] - 3 - nitrobenzene - 1 - sulfonyl} - 4, 5, 6, 7 - tetrahydrothieno [3, 2 - *c]* pyridine (2-a).

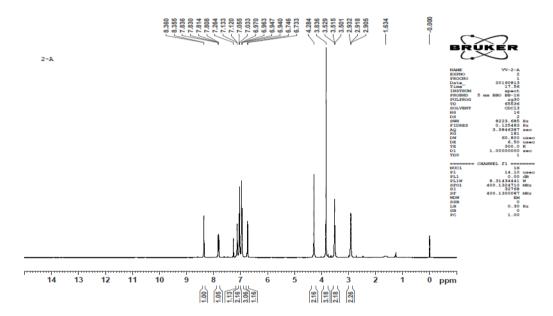


Figure 2.

Table 3: ¹H NMR data of compound 2a.

Chemical No of Pro-		Multiplicity and	
Shift	ton	J value in Hz	
8.360-8.355	1	Doublet J=2	
7.836-7.808	1	Doublet of doublet J=2.4, 8.8	
7.133-7.120	1	Doublet J=5.2	
7.055-6.940	4	Multiplet	
6.746-6.733	1	Doublet J=5.2	
4.225	2	Singlet	
3.466-3.437	2	Triplet J=5.6, 11.6	
2.934-2.907	2	Triplet J=5.6, 10.	
2.041-2.026	4	Multiplet J=6	

¹³C NMR Spectral study of 5 - {4 - [(4 - methoxyphenyl) sulfanyl] - 3 - nitrobenzene - 1 - sulfonyl} - 4, 5, 6, 7 - tetrahydrothieno [3, 2 - *c*] pyridine (2-a).

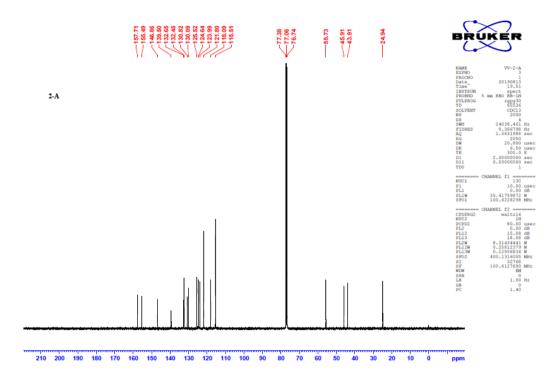


Figure 3.

Table 4: ¹³CMR data of compound 2a.

δ Value	No. Of Carbon
24.94	1C
43.91	1C
45.91	1C
55.73	1C
115.51	1C
118.09	1C
121.80	1C
123.99	1C
124.64	1C
125.52	2C
130.09	1C
130.82	1C
132.45	1C
132.65	2C
139.50	1C
146.86	1C
155.49	1C
157.71	1C

Mass spectroscopic study of $5 - \{4 - [(4 - methoxyphenyl) sulfanyl] - 3 - nitrobenzene - 1 - sulfonyl\} - 4, 5, 6, 7 - tetrahydrothieno <math>[3, 2 - c]$ pyridine (2-a).

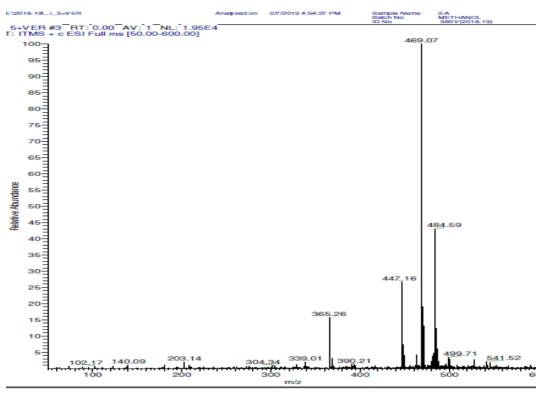


Figure 4.

Mass in m/z

Molecular ion peak was observed at 447.16 [M + 1]

ANTIMICROBIAL ACTIVITY

The MICs of all the synthesized compounds was carried out by broth dilution method. Mueller Hinton broth was used as nutrient medium to grow and dilute the compound suspension for the test bacteria and Sabouraud Dextrose broth was used for fungal nutrition. Inoculum size for test strain was adjusted to 10⁸ CFU [Colony Forming Unit] per milliliter by comparing the turbidity. Dimethyl sulfoxide (DMSO) was used as diluent to get desired concentration of drugs to test on standard bacterial strains. Serial dilutions were prepared in primary and secondary screening. The control tube containing no antibiotic was immediately subcultured (before inoculation) by spreading a loopful evenly over a quarter of plate of medium suitable for the growth of the test organism and put for incubation at 37 °C overnight. The tubes were then incubated overnight. The MIC of the control organism was read to check the accuracy of the drug concentrations. The lowest concentration inhibiting growth of the organism was recorded as the MIC. All the tubes not showing visible growth (in the same manner as control tube described above) was subcultured and incubated overnight at 37 °C. The amount of growth from the control tube before incubation (which represents the original

inoculum) was compared. Subcultures might show similar number of colonies indicating bacteriostatic; a reduced number of colonies indicating a partial or slow bactericidal activity and no growth if the whole inoculum has been killed. The test must include a second set of the same dilutions inoculated with an organism of known sensitivity. Each synthesized drug was diluted to $2000~\mu g/ml$ concentration, as a stock solution. In primary screening 1000, 500 and $250~\mu g/ml$ concentrations of the synthesized drugs were taken. The active synthesized drugs found in this primary screening were further tested in a second set of dilution against all microorganisms. The drugs found active in primary screening were similarly diluted to obtain 200, 125, 100, 62.5, 50 and $25~\mu g/ml$ concentrations. The highest dilution showing at least 99% inhibition is taken as MIC. The activities of all the synthesized compounds are recorded in Table 1.

Table 1: Antimicrobial activity of 5 - $\{4 - [(4 - methoxyphenyl) sulfanyl] - 3 - nitrobenzene - 1 - sulfonyl\} - 4, 5, 6, 7 - tetrahydrothieno <math>[3, 2 - c]$ pyridine.

Table 5 Antimicrobial activity of 5 –[4- (4 - Substituted phenoxy - 3- Nitrobenzene - 1 - Sulfonyl) - 4, 5, 6, 7 –Tetrahydrothieno [3, 2 - C] Pyridine.

	Minimal bactericidal concentration (μg/ml)				Minimal fungicidal
Comp. No.	Gram-negative		Gram-positive		Concentration (µg/ml)
	E. coli	P. aeruginosa	S. aureus	S. pyogenes	A. niger
2a	150	100	100	200	250
2b	200	200	200	150	1000
2c	125	150	150	100	500
2d	150	125	100	150	1000
Ampicillin	100		250	100	
Chloramphenicol	50	50	50	50	
Nystatin					100
Griseofulvin					100

ANTIMICROBIAL EVALUATION

In the series 2a-d, compound 2c and 2d found comparative inhibitory activity against E.coli. with respect to ampicillin. Compound 2c and 2d having electron withdrawing fluorine and chlorine group at ortho and para position respectively found two fold lessactive against E. coli. with respect to ampicillin. Compound 2b having electron withdrawing group cyno at para position of the phenyl ring found more than two fold less active against E. coli. with respect to ampicillin. Compound 2a having electron donaing methoxy group at para position of the phenyl ring respectively found almost two fold less potent against E. coli. with respect to ampicillin.

For p. aeruginosa, compound 2a having electron donating methoxy group at para position found most reactive in the series, however, chloramphenicol was still two fold more potent than compound 2a. Compound 2b, 2c and 2d having electron withdrawing group cyno, chloro and fluoro at ortho or para position found to possess good inhibitory activity in the series, still almost two fold less active compare to chloramphenicol. Replacement of electron withdrawing group ortho to para position of the phenyl ring respectively found more active but still they found respectively less active against p.aeruginosa with respect to chloramphenicol.

For s.aureus, all the compound (2a-d) found more potent against s. aureus with respect to ampicillin. Compound 2d having electron withdrawing chlorine and fluorine at ortho or para position respectively found more than two fold more active against s. aureus with respect to ampicillin. Compound 2a having electron donating methoxy group at para position respectively found more than two fold more reactive against s. aureus with respect to ampicillin. Compound 2b having electron withdrawing cyno group at para position decrease the inhibition activity by almost two fold, however, still it found more potent against s. aureus with respect to ampicillin.

For s. pyogenes, compound 2c having electron withdrawing group at ortho position of the phenyl ring found equipotent against s. pyogenes with respect to ampicillin. All other compound in these series found less potent against s. pyogenes with respect to ampicillin. These finding suggested that phenols having electron withdrawing group at ortho position of the phenyl ring are better substituent in chloro nitro pyridine.

For fungi, compound 2a having electron donating methoxy group at para position of the phenyl ring found most potent against among the series (2a-d), however, found still more than two fold less potent against A. niger with respect to nystatin and griseofulvin.

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