

AN OVERVIEW ON MICROWAVE ASSISTED SYNTHESIS OF ANTIMALARIAL COMPOUNDS

Abhishek Chaudhary, Divya Arora* and Pooja Devi

Abhilashi University, Chail Chowk, Tehsil Chachyot, Mandi, Himachal Pradesh- 175045,
India.

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*Corresponding Author

Divya Arora

Abhilashi University, Chail
Chowk, Tehsil
Chachyot, Mandi, Himachal
Pradesh- 175045, India.

ABSTRACT

The present review article offers a detailed account of the microwave irradiation synthesis of novel ligands for antimalarial potential. Microwave synthesis being an economical method, time saving and environmental friendly will be best suitable for future of antimalarial drugs. The synthesis by microwave irradiation and antimalarial studies of various drugs as per the future need of medical and pharmaceuticals have been discussed. Some of the derivatives synthesised by microwave irradiation method discussed here includes: benzothiazole-spirooxindole derivatives; pyrimidine derivatives; 4-benzyloxy- and 4-aryloxy-2 trichloromethylquinazolines; 2, N₆-disubstituted 1,2-dihydro-1,3,5-triazine-4,6-diamines; 4-aminoquinolines; etc.

KEYWORDS: Malaria, Microwave irradiation, Antimalarial potential.

INTRODUCTION

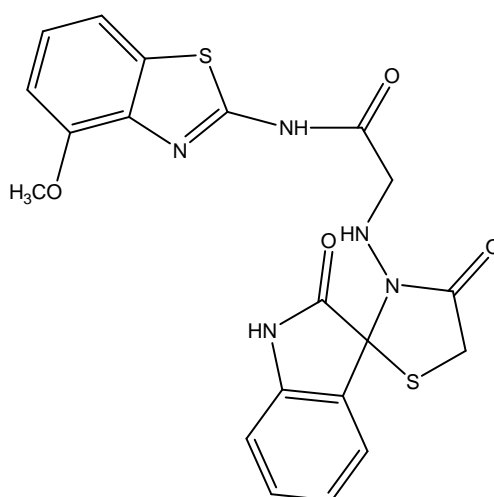
Malaria is a major health concern in many tropical and subtropical nations. Over the last fifteen years, further efforts in the development of novel antimalarials have lowered the frequency of this global illness.^[1] *Plasmodium falciparum*, *Plasmodium Vivax*, *Plasmodium Ovale*, and *Plasmodium Malariae* are all *Plasmodium* species that pose a serious health risk to humans. *Plasmodium falciparum* considered most threat full in terms of death and *Plasmodium Vivax* considered major cause of illness and death across large parts of the world.^[2] The simian parasite *Plasmodial knowlesi* has lately emerged as a major cause of malaria in Malaysia and Southeast Asia. There is no evidence of human-to-human transmission of this parasite.^[3]

Green chemistry is a new and fast growing alternative to standard chemical synthesis methods. Even after maximal resource use, it refers to the generation of the smallest amount of chemical waste possible. Green chemistry prevents the creation of by-products as well as the usage of toxic and dangerous solvents. As a result, they are adaptable for use in chemistry's combinatorial synthesis.^[4] In 1986, Gedye and Giguere proposed using microwaves to speed up organic chemistry reactions. Microwave irradiation was thus introduced into organic chemical synthesis.^[5] Because of the quick reaction time and rapid optimization of chemical reactions, it became a popular technology.^[6-8] This method is gaining popularity as a means of speeding up the drug research and development process.

REVIEW

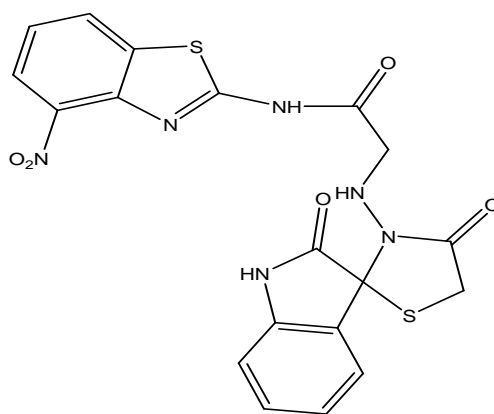
a. Benzothiazole Spiro-Oxindole Derivatives

Under microwave irradiation, novel benzothiazole-spirooxindole compounds were synthesised from 2-aminobenzothiazoles and spirooxindoles, which were studied using several spectroscopic techniques. The *Plasmodium falciparum* strain was tested in vitro for antimalarial activity.^[9] Almost all of the compounds had good antimalarial action, therefore Chloroquine and Quinine remained the standard medications. Compounds 1 and 2 had MIC values (0.082 and 0.049 g/mL) that were nearly identical to the standard drug. Chloroquine, compound 3 and 4, and compound 3 and 4 had lower MIC values (0.32 and 0.38 g/mL) than Quinine. The synthesised benzothiazole-spirooxindole compounds will aid in the development of innovative antimalarial drugs in the future.^[10]

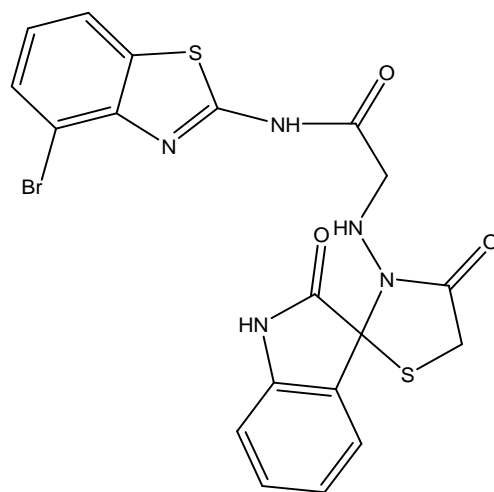


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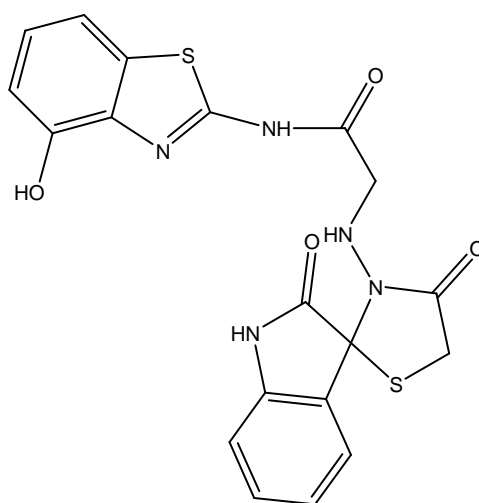
MIC value = 0.082 μ g/mL



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MIC value = 0.049 $\mu\text{g/mL}$ 

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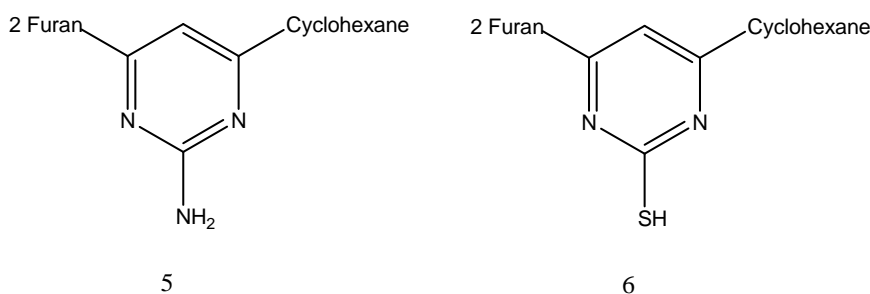
MIC value = 0.32 $\mu\text{g/mL}$ 

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MIC value = 0.38 $\mu\text{g/mL}$

b. Novel Pyrimidine Derivatives

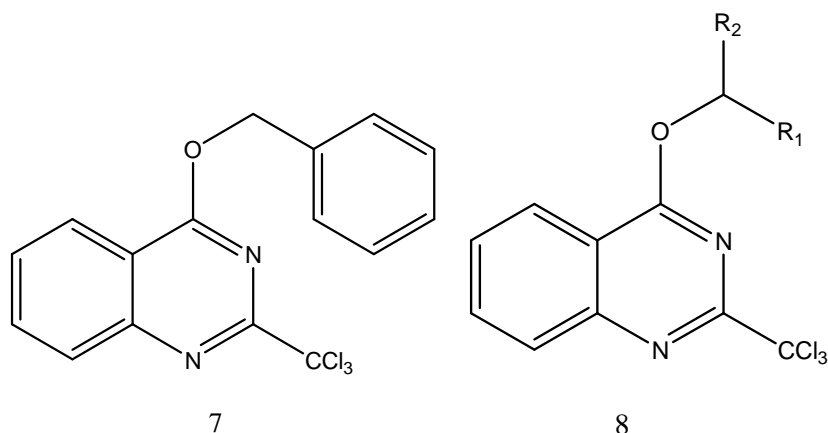
The new pyrimidine derivatives were made from chalcones and tested in vitro for antimalarial activity against *Plasmodium falciparum* strain cells. The reference medications were quinine and chloroquine. All of the chalcone compounds were shown to be ineffective against *Plasmodium falciparum*, however synthesised pyrimidine derivatives were highly effective. Compounds 5 and 6, with MIC values of 0.028 and 0.078 g/mL, respectively, showed good antimalarial activity among all produced derivatives. The bioassay revealed that pyrimidines have higher bioactivity than their chalcone counterparts, indicating that the pyrimidine ring is responsible for the increase in activity. As a result, amino-pyrimidines appeared to be more potential for future development.^[11]



MIC value 0.028 and 0.078 µg/ml, respectively

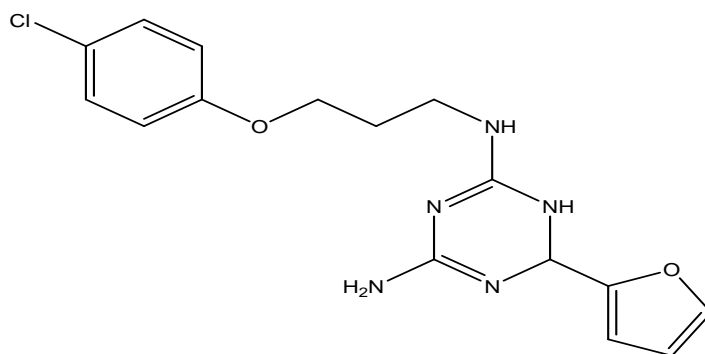
c. 4-benzyloxy- and 4-aryloxy-2- trichloromethylquinazolines

An aniline^[12], aryl^[13], thiophenol^[14], alkynyl^[15], or sulphonamide moiety^[16] was substituted at position 4 of the quinazoline ring to produce a range of compounds. A 4-phenoxy-2-trichloromethyl quinazoline derivative with an IC₅₀ of 1.1 M, a CC₅₀ of 50 M, and a selectivity index of 45 was also discovered^[17], using Chloroquine and Doxycycline as standards. The multi-resistant *Plasmodium falciparum* strain K1 was used to investigate in vitro activity. The synthesis was carried out utilising the S_NAr reaction, which was carried out with benzylic alcohols, phenols, or hydroxyheterocycles under microwave irradiation. This approach for substituting an alkoxy, phenoxy, or heteroaryloxy moiety at position 4 of 4-chloroquinazolines was efficient, simple, rapid, and inexpensive.^[18]



d. 2,N6-disubstituted-1,2-dihydro-1,3,5-triazine-4,6-diamines

In a variety of cycloguanil-like compounds, the flexible tether length was interpolated between the 1,2-dihydro-1,3,5-triazine-4,6-diamine heterocycle and the substituted phenyl ring.^[19] Microwave synthesis, as a one-pot approach, allowing the conversion to be completed in 1-2 hours.^[20-22] Because of the Asn51Ile and Ile164Leu mutations, which create a chain shift around the Plasmodium falciparum DHFR active site, minor inhibitors like cycloguanil and pyrimethamine have a decreased binding affinity. Larger substituents at C-2 on the 1,2-dihydro-1,3,5-triazine-4,6-diamine ring were introduced to accommodate the increased binding site available. Examples include cycloalkyl, phenyl, and substituted phenyl. The FCR-3 Plasmodium falciparum strain was used to test a total of 28 drugs. The majority of the compounds had antimalarial activity comparable to cycloguanil ($IC_{50} = 4.995 \text{ M}$), with the exception of a molecule containing a 2-furanyl derivative, which was about 5-fold more active ($IC_{50} = 0.99 \text{ M}$) than cycloguanil. The absence of haemolysis and red blood cell toxicity showed that the compounds had direct efficacy against the intra-erythrocytic parasite.^[23]

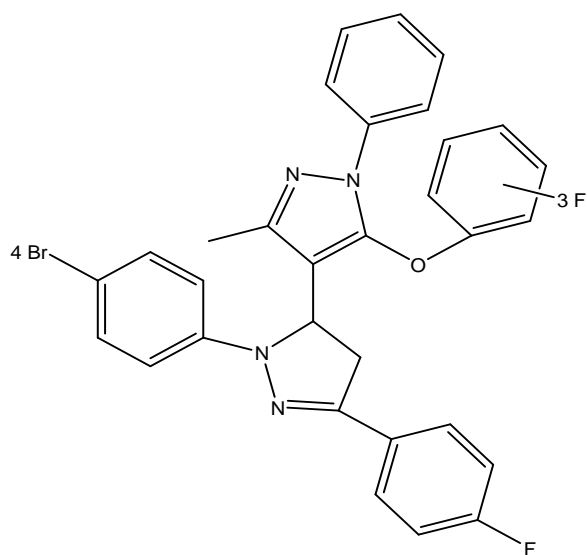


IC_{50} value = $0.99 \mu\text{M}$

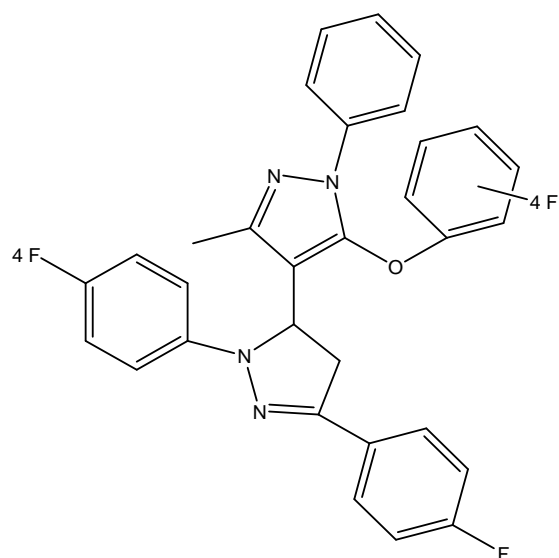
In the context of the synthesis of physiologically active new heterocyclic scaffolds, a novel microwave assisted synthesis of fluorinated pyrazolylpyrazoline derivatives was reported.^{[24-}

All of the compounds were evaluated for antimalarial activity against quinine and chloroquine-resistant *Plasmodium falciparum* isolates. The mean IC₅₀ for the *Plasmodium falciparum* strain was determined to be between 0.022 and 0.088 M. Certain chemicals showed excellent action against the *Plasmodium falciparum* strain as compared to quinine, which has an IC₅₀ of 0.268 M. As a result, these compounds may one day prove to be novel antimalarial agents.^[32]

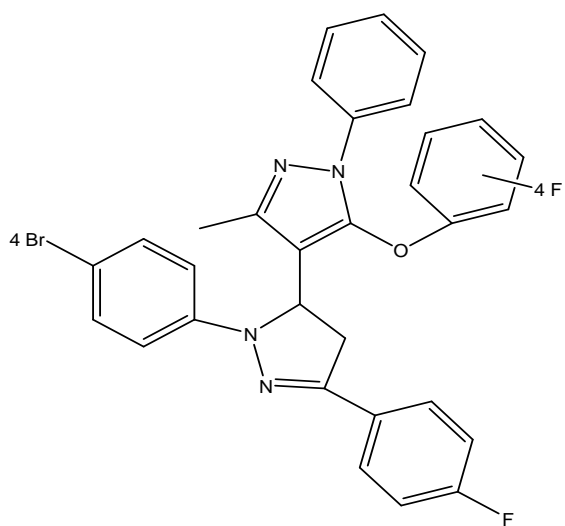




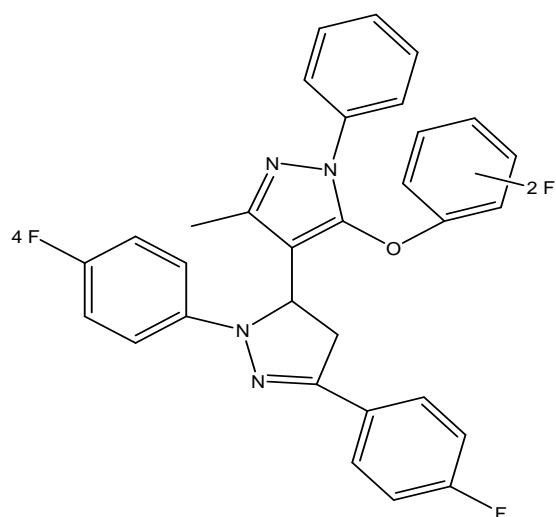
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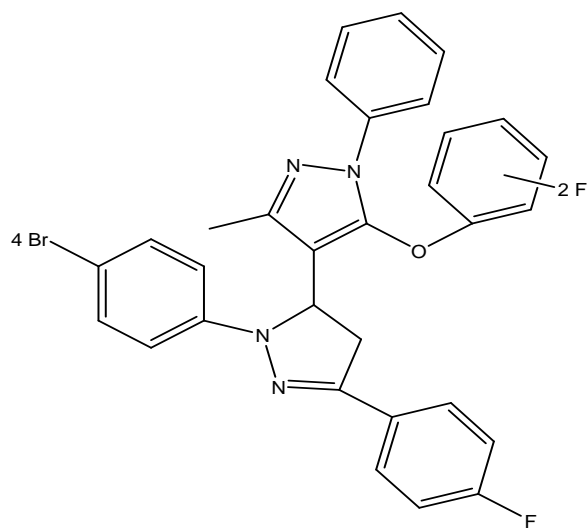
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CONCLUSION

As per the future need of the environment the microwave irradiation is the most required method for any synthesis. The method is very time saving, environment friendly and economical as well. Many antimalarials were developed against the *Plasmodium* species, but due to development of resistance for the compound synthesized, there is dire need for the new novel antimalarial compounds. Also, environment is the basic topic of concern and use of benign chemicals for synthesis of antimalarials is required. Here, this review covers some compounds that can be synthesized in microwave by use of environmentally benign chemicals.

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